

References and Notes

1. T. Ewing, J. A. Grand, R. R. Miller, *J. Am. Chem. Soc.* **74**, 11 (1952); *J. Phys. Chem.* **58**, 1086 (1954).
 2. T. Ewing, J. P. Stone, J. R. Spann, E. W. Steinkuller, D. D. Williams, R. R. Miller, "High Temperature Properties of Sodium and Potassium," *12th Progress Report for U.S. Naval Research Laboratory, NRL Rept. 6094* (Washington, D.C., 9 June 1964).
 3. A. W. Lemmon, Jr., H. W. Deem, E. A. Aldridge, E. H. Hall, J. Matolich, Jr., J. F. Walling, "Engineering Properties of Potassium," *NASA CR-54017, BATT-4673-Final* (Battelle Institute, Columbus, Ohio, 1963).
 4. I. I. Novikov *et al.*, *J. Nucl. Energy* **4**, 387 (1957).
 5. A. V. Grosse, *Science* **140**, 784 (Table 4) (1963).
 6. ———, *J. Phys. Chem.* **68**, 3419 (1964).
 7. ———, *J. Inorg. Nucl. Chem.* **22**, 23 (1961).
 8. E. N. da C. Andrade, *Phil. Mag.* **17**, 698 (1934); both his I and II equations are also fully discussed in ref. 7.
 9. M. Sittig, "Sodium, Its Manufacture, Preparation and Uses," *Am. Chem. Soc. Monogr. No. 133* (Reinhold, New York, 1956), pp. 456-461.
 10. J. O. Hirschfelder, C. F. Curtiss, R. B. Bird, *Molecular Theory of Gases and Liquids* (Wiley, New York, 1954), p. 14.
 11. L. Pauling, *The Nature of the Chemical Bond* (Cornell Univ. Press, Ithaca, N.Y., ed. 3, 1960), p. 403.
 12. For mercury, the constant is 83×10^{-4} ; thus it may change to some extent in the various families of the periodic system.
 13. O. A. Hougen and K. M. Watson, *Chemical Process Principles* (Wiley, New York, 1943), vol. 3, p. 873.
 14. A. V. Grosse, *Inorg. Chem.* **1**, 436 (1962).
 15. This work was supported by AEC contract AT(30-1)-2082.
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Superconducting Gallium Antimonide

Abstract. A metallic phase of gallium antimonide, obtained by quenching at approximately 120 kilobars to 77°K and then releasing pressure, is a superconductor. The transition temperature depends on the annealing conditions; for samples annealed at 250°C under pressure before quenching, it is $4.24^\circ \pm 0.10^\circ\text{K}$, and H_{c2} (the critical field) equals 2640 gauss at 3.50°K. This temperature is higher than the 2.1°K reported for metallic indium antimonide.

Gallium antimonide transforms from a semiconducting to a metallic state at approximately 70 kb at 25°C (1). X-ray powder photographs taken at high pressures show that the metallic phase has a structure similar to that of white tin (2). Recently the high-pressure phase has been retained by quenching the sample to 77°K before releasing pressure (3, 4). X-ray powder photographs taken at 77°K and 1 bar confirm the "white tin" structure and indicate that the high-pressure phase has been retained (3). Because the metallic phase of indium antimonide is superconducting at 2.1°K (5), we have determined the superconducting properties of metallic gallium antimonide in order to compare them with those of InSb.

Three different sources of gallium antimonide were used: (i) single-crystal material from Merck and Co., (ii) single-crystal GaSb doped with about 0.01 percent Te to reduce the possibility of excess gallium (6), and (iii) p-type polycrystalline material from American Smelting and Refining Co. (Asarco). The samples were compressed between tungsten carbide anvils with 2.4-mm faces to an average pressure of 120 kb and then cooled to 77°K before pressure was released; some were annealed by heating the anvils before quenching. The resulting samples were discs ~ 0.05 mm thick and 2.5 mm in diameter. The samples were transferred at 77 K to a helium cryostat and tested

for superconductivity by the alternating-current method (7).

Superconducting properties of the quenched gallium antimonide apparently depend on the annealing conditions (Table 1). The transition temperature, T_c , and the hardness (that is, the relative strength of the magnetic field necessary to destroy the superconducting state) vary differently with annealing. Annealing at temperatures above 100°C causes T_c to drop from 5.9° toward 4.2°K; annealing at 50°C appears to make the samples magnetically softer. An annealed and an unannealed sample were reconverted by warming them to room temperature. X-ray powder photographs of the re-

converted material showed two broad halos centered around the first few lines of the zinc-blende structure. These samples were tested for superconductivity, and the negative result indicated that the retained phase is responsible for the observed superconductivity. X-ray diffraction patterns taken at 77°K of annealed (200°C) and unannealed samples show the "white tin" structure. The data were not sufficiently accurate to determine whether there was a small systematic variation in lattice parameter with annealing. The samples annealed at 200°C also showed two faint additional lines at 3.3 and 1.65 Å. These lines may be attributable to small amounts of GaSbO₄, but when an annealed sample was reconverted by heating for 1 hour at 200°C an x-ray photograph showed only the diffraction lines of the zinc-blende form of GaSb. If the additional lines at 3.3 and 1.65 Å were from GaSbO₄ they should have appeared in the film of the reconverted material.

The large change in T_c for GaSb observed on annealing may result from an order-disorder transition or from the relief of strains in the sample. The extra faint lines observed in the annealed samples may be superstructure lines; for example, the 111 and 222 reflections for a cell with the c-axis of the "white tin" structure doubled. However, available data do not permit an unambiguous description.

It is also known that strain can substantially change the transition temperature. For example, a difference of > 1°K was observed in annealed and unannealed specimens of Nb containing 10 percent Cr (8). There was similar change in samples of InTe quenched from high pressures and tem-

Table 1. Superconductivity of GaSb; eight independent experiments.

Sample	T_c (°K)	H (gauss)	Annealing		Source
			Temp. (°C)	Time (min)	
2544	4.20-4.28 3.46	0 2640	250	60	Merck
2522	4.24-4.38 3.82 3.50	0 1640 2640	200	90	Te "doped"
2530	4.24-4.38 4.12	0 660	200	15	Merck
2529	4.45-4.90 (4.2 ± 0.5)*	0 (1000)	100	45	Merck
2534	5.85-6.05 (3 ± 1)*	0 (4600)	50	30	Asarco
2519	5.85-6.15 5.44	0 4600			Te doped
2523	5.75-6.05	0			Merck
2507	5.40-6.15	0			Merck

* Obtained by cycling the magnetic field at about constant temperature.